

Reaction of α -Vinylallenylphosphonates with Diazoalkanes

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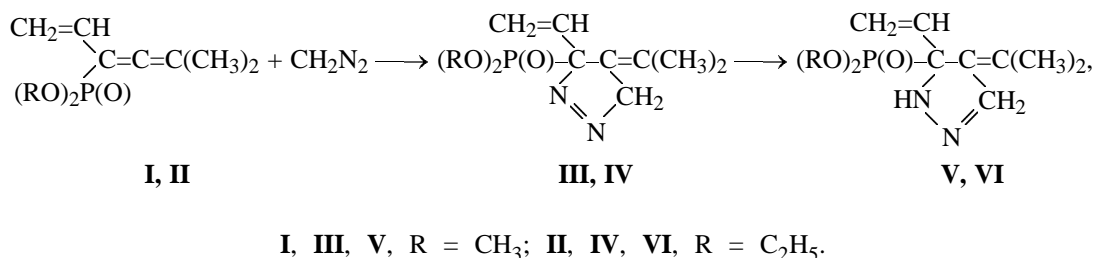
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Abstract—Diazomethane reacts with 3-methyl-1-vinyl-1,2-butadienylphosphonates following the scheme of [3+2]-cycloaddition at the 1,2-double bond of the allenic fragment. The reaction with diphenyldiazomethane involves the vinyl fragment and yields substituted diphenylcyclopropane.

Phosphorylated allenes are known to react with diazoalkanes at the 1,2-double bond according to the [3+2]-cycloaddition pattern [1, 2]. Introduction of a vinyl group into the α -position with respect to the phosphinoyl fragment gives rise to additional unsaturated reaction centers in the dipolarophile: apart from the cumulene triad, conjugated 1,3-diene system and an ethylene group appear.

We have studied addition of diazomethane and diphenyldiazomethane to dimethyl and diethyl 3-methyl-1-vinyl-1,2-butadienylphosphonates **I** and **II**. The IR spectra of the resulting 1:1 adducts lack stretching vibration band typical of allene triad (1955–

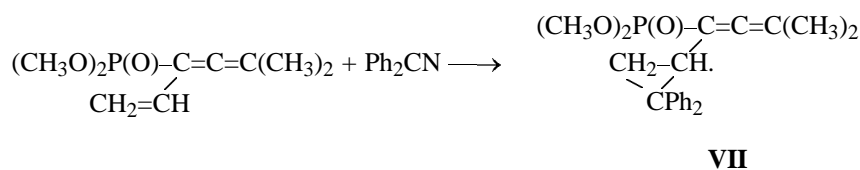
1960 cm^{-1}), indicating that the vinyl group is not involved. The presence of IR absorption at 900 cm^{-1} , typical of out-of-plane bending vibrations of the $=\text{CH}_2$ group [3] suggests that the diene system is also retained. The same follows from the ^1H NMR spectra of the adducts, which contain multiplets at δ 4.8–5.6 and 6.1–6.5 ppm, typical of vinyl group protons. The presence of an isopropylidene fragment is supported by characteristic signals from methyl groups at a double bond, δ 1.8 ppm. These data led us to conclude that the role of dipolarophile in the reaction with diazomethane is played by the 1,2-double bond of the cumulene triad.



Freshly synthesized adducts **III** and **IV** show in the IR spectra a weak absorption at 3400 (NH) and 1560 cm^{-1} ($\text{C}=\text{N}$), whose intensity increases with rise in temperature or on dissolution in proton-donor solvents. Presumably, this is the result of tautomerization of primary products, 4,5-dihydro-3H-pyrazoles **III** and **IV**, into the corresponding 1H-isomers **V** and

VI. In the ^1H NMR spectra, an olefinic proton signal appears at δ 6.9 ppm (3-H).

Phosphorylated triene **I** reacts with diphenyldiazomethane less vigorously. Analysis of the IR and ^1H NMR spectra of the adduct indicates that the reaction involves the vinyl group of **I**.



The IR spectrum of **VII** contains a strong absorption band at 1960 cm^{-1} from the cumulene system. In the ^1H NMR spectrum of adduct **VII**, cyclopropane ring protons give a triplet at δ 2.18 ppm (CH) and a doublet at δ 1.38 ppm (CH_2). Signals from the aromatic protons are observed in the region δ 7.0–7.44 ppm.

EXPERIMENTAL

The IR spectra were recorded on a UR-20 spectrometer. The ^1H NMR spectra were obtained on a Varian T-60 instrument (60 MHz) in carbon tetrachloride using tetramethylsilane as reference.

3-Dialkoxyphosphinoyl-4-isopropylidene-3-vinyl-4,5-dihydro-3H-pyrazoles III and IV. A mixture of 0.01 mol of phosphonate **I** or **II** and 0.015 mol of diazomethane in ether was left to stand for a week at 5°C . The solvent was removed to leave a viscous oily substance.

3-Dimethoxyphosphinoyl-4-isopropylidene-3-vinyl-4,5-dihydro-3H-pyrazole (III). Yield 2.3 g (96%), n_{D}^{20} 1.5370. Found, %: C 50.01; H 6.85; P 12.50. $\text{C}_{10}\text{H}_{17}\text{N}_2\text{O}_3\text{P}$. Calculated, %: C 49.18; H 6.97; P 12.71.

3-Diethoxyphosphinoyl-4-isopropylidene-3-vinyl-4,5-dihydro-3H-pyrazole (III). Yield 2.6 g

(97%), n_{D}^{20} 1.5128. Found, %: C 53.12; H 8.09; P 10.91. $\text{C}_{12}\text{H}_{21}\text{N}_2\text{O}_3\text{P}$. Calculated, %: C 52.94; H 7.72; P 11.40.

1-Dimethoxyphosphinoyl-1-(2,2-diphenylcyclopropyl)-3-methyl-1,2-butadiene (VII). A mixture of 1.94 g of diphenyldiazomethane and 2 g of phosphonate **I** in 10 ml of petroleum ether was left to stand for 2 weeks at 15°C . The crystals were filtered off and washed with ether. Yield 2.3 g (64%), mp $94\text{--}96^\circ\text{C}$. Found, %: C 72.11; H 7.04; P 7.85. $\text{C}_{22}\text{H}_{25}\text{O}_3\text{P}$. Calculated, %: C 71.74; H 6.79; P 8.42.

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